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V.P. & R.P.T.P.SCIENCE COLLEGE

(SEMESTER – VI) INTERNAL EXAMINATION

Physical	Chemistry: USU6CCHE05 Time: 11:00 a.m. to 12:30 p.m.	
Date: 16-	03-2018, Friday Total Marks: 25	
Q-1:	Choose the correct option from the following.(Multiple choice question)	[03]
(i)	The m-dichlorobenzene has dipole moment o-dichlorobenzene.	
	(a) same as (b) greater than (c) less than (d) none of the above	
(iii)	A silver iodide sol was prepared by mixing AgNO ₃ and KI solution with KI in slight excess. The AgI sol is (a) negatively charged because of excess I adsorbed (b) positively charged because of excess K ⁺ adsorbed (c) negatively charged because of excess NO ₃ adsorbed (d) neutral The rotational degree of freedom for non-linear molecule is	Collegi
	(a) 1 (b) 2 (c) 3 (d) none	*
Q-2:	Answer the following. (Any two)	[04]
(i)	What are the conditions required to observe vibrational spectra.	
(ii)	Define: (a) Peptization (b) Coagulation	
(iii)	Give the Clausius-Mosotti equation for polar molecules. State the each terms involved.	
Q-3	Derive an expression for determination of molecular parameters of diatomic polar molecule from pure rotational spectra.	[06]
	OR	
Q-3 (a)	Describe various modes of vibration of polyatomic molecule giving suitable example.	[03]
(b)	Sketch and explain P-Q-R bands observed in IR spectra.	[03]
Q-4 (a)	Describe the Vapour-Temperature method for measuring the dipole moment of a molecule.	[03]
(b)	Describe the various types of induced polarization of molecules in an electric field.	[03]
	OR	
Q-4 (a)	Describe the principle, construction and working of Abbe's refractometer.	[03]

The bond length of H-F bond is 0.9168 Å and dipole moment observed is 1.909 D. [03] (b) Calculate the ionic bond character of HF molecule. Q-5 (a) [03] Discuss the methods for the purification of colloidal solutions. Distinguish between true solution, a colloidal solution and a suspension. [03] (b) OR Q-5 (a) Distinguish between Lyophilic Sols and Lyophobic Sols. [03] P. Scien (b) Explain four important applications of colloids. [03]