V. P. & R. P. T. P. SCIENCE COLLEGE, V. V. NAGAR. INTERNAL TEST: MARCH-2018

T. Y. B. Sc. Semester-VI Sub.:- Inorganic Chemistry, Course Code :-US06CCHE04 Date: 15/03/2018 Total Marks:25	
Note: (i) All questions are to be attempted.	
(ii) Figures to the right of each question indicate full marks.	
Q: 1Give the most correct choice to the following multiple choice	
questions.	[3]
(i) Although, is placed near top of the electro-motive	
force series, yet it resists atmospheric corrosion effectively.	P. Scie
(a) zinc (b) aluminium (c) sodium (d) copper	(Q.
(ii) is not effect of alloying.	LIBRAF
(a) Decreasing melting point (b) Increasing hardness of metal	1/2
(c) Decreasing castability (d) Modifying colour	V. Nac
(iii) Mark the compound which gives carbon with concentrated H ₂ SO	4.
(a) Starch (b) Oxalic acid (c) Ethanol (d) Formic acid	
Q: 2 Answers the following short questions(any two).	[4]
(i) Explain immersed corrosion by "acid-theory."	
(ii) List different types of steel and explain any one of them.	
(iii)Explain, sulphuric acid neutralizes alkalis to give two series of sa	ilts.
Q:3[A]Explain the term 'passivity' and discuss the protective layer theor	ry. [3]
[B]Discuss the prevention of corrosion by producing insoluble	
oxide coating and by hot dipping.	[3]
OR	
Q:3[A]Describe the factors determining rate of corrosion reaction for	
metal sheltered from rain in gaseous environments.	[3]
[B]Corrosion starts from metal joints. Explain.	[3]
Q: 4[A]Give the classification of alloys. Discuss non-ferrous alloys	
with suitable examples.	[3]
[B]Write note on: Inter-metallic compounds.	[3]

OR

Q: 4[A]Discuss in detail the fusion method for the preparation alloys. [B]Explain substitutional alloys and discuss the phase diagram	[3]
of brass alloy.	[3]
Q: 5[A] Explain nitric acid as an important oxidizing reagent.	[3]
[B]Describe the manufacture of caustic-soda by using Nelson cell.	[3]
OR	
Q:5[A]Describe the contact process for the manufacture of sulphuric ac	id. [3]
[B]Discuss the manufacture of nitric acid by Ostwald's process	
in detail.	[3]
