SUBJECT : ORGANIC CHEMISTRY (US05CCHE01)
TIME : 11.00 a.m. to 12.30 p.m. DATE : 29-09-2016 TOTAL MARKS : 25
Q. 1 Choose the correct option for the following
(i) Which of the following compound have the properties of secondary aliphatic amine?
(a) Pyridine
(b) Piperidine
(c) Methyl amine
(d) Pyrrole.
(ii) How many NMR signals would you expect from allyl alcohol?
(a) 3
(b) 4
(c) 5
(d) 6 .
(iii) Which of the following is the example of co-polymer?
(a) PVC
(b) Plexiglas
(c) Orlon
(d) SBR.
Q. 2 ANSWER THE FOLLOWING (ANY TWO)
(i) Give the synthesis of $\beta$-aminopyridine from $\beta$-picoline.
(ii) Give the various aspects of CMR spectroscopy.
(III) Write the chemical structure of following synthetic polymer with their most likely monomer.
(a) Saran
(b) Carbowax.
Q. 3 ANSWER THE FOLLOWING
(i) Discuss why nucleophilic substitution reaction in pyridine is preferred at the 2 - and 4-positions.
(ii) Arrange the following compounds in increasing basicity order and give detail explanation of your answer.
(a) Imine
(b) Amine
(c) Nitrile.
OR
Q. 3 ANSWER THE FOLLOWING
(i) Why electrophilic substitution reaction in pyrrole occurs at carbon atom and not on hetero atom.
(ii) Give the synthesis of 5,6-Benzoquinoline from 2-aminonaphthalene and glycerol by Skraup synthetic route.
Q. 4 ANSWER THE FOLLOWING
(i) Molecular formula: $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{O}$

CMR ( $\delta, \mathrm{ppm}$ ) :
(a) 3.4, Quartet
(b) 50.8, Triplet
(c) 77.9 , Singlet
(d) 81.6, Singlet.

NMR ( $\delta, \mathrm{ppm}$ ) :
(a) $2.0,3 \mathrm{H}$, Singlet
(b) $1.8,1 \mathrm{H}$, Singlet
(c) $4.1,2 \mathrm{H}$, Singlet.
(ii) Molecular formula: $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{O}_{2}$

NMR ( $\delta, \mathrm{ppm}$ ) : (a) $7.5,4 \mathrm{H}$, Quartet (b) $3.9,3 \mathrm{H}$, Singlet (c) $2.5,3 \mathrm{H}$, Singlet.
IR (Cm $\left.{ }^{-1}\right): 3000,2900,1670,1598,1500,1450,1375,1258,1171,1021,833$.
OR
Q. 4 ANSWEP THE FOLLOWING
(i) Molecular formula: $\mathrm{C}_{9} \mathrm{H}_{10}$
$\operatorname{IR}\left(\mathrm{CM}^{-1}\right): 3100,2950,1650,1600,1500,1450,1375,890,760-770$.
NMR ( $\delta, \mathrm{ppm}$ ) :
(a) 7.4,5H, Complex
(b) $5.35,1 \mathrm{H}$, Singlet
(c) $5.1,1 \mathrm{H}$, Singlet
(d) 2.10, 3H, Singlet.
(ii) Discuss the proton-coupled and proton-decoupled CMR spectrum of sec-butyl bromide.
Q. 5 ANSWER THE FOLLOWING
(i) What are dienes ? Just classify the following dienes into appropriate class.
(a) 2,4-hexadiene
(b) Allene
(c) 1,4-pentadiene.
(ii) What is coordination polymerization ? Explain the importance of Ziegler-Natta catalyst in coordination polymerization.

## OR

Q. 5 ANSWER THE FOLLOWING
(i) Discuss the addition of HBr to 1,3 -butadiene at $-80^{\circ} \mathrm{C}$ and at $40^{\circ} \mathrm{C}$ temperature with potential energy diagram.
(ii) Give the distinguishing features of addition polymerization and condensation polymerization.

There is no short cut, except hard work with understanding to excel in examination.


