# V. P. AND R. P. T. P. SCIENCE COLLEGE VALLABH VIDYANAGAR B.Sc. (SEMESTER-V) INTERNAL TEST-2016 SUBJECT : ORGANIC CHEMISTRY (US05CCHE01) TIME : 11.00 a.m. to 12.30 p.m. DATE : 29-09-2016 TOTAL MARKS : 25

### Q. 1 Choose the correct option for the following

- (i) Which of the following compound have the properties of secondary aliphatic amine ?
  - (a) Pyridine (b) Piperidine (c) Methyl amine (d) Pyrrole.
- (ii) How many NMR signals would you expect from allyl alcohol?
  - (a) 3 (b) 4 (c) 5 (d) 6.
- (iii) Which of the following is the example of co-polymer?
  - (a) PVC (b) Plexiglas (c) Orlon (d) SBR.

## Q. 2 ANSWER THE FOLLOWING (ANY TWO)

- (i) Give the synthesis of  $\beta$ -aminopyridine from  $\beta$ -picoline.
- (ii) Give the various aspects of CMR spectroscopy.
- (III) Write the chemical structure of following synthetic polymer with their most likely monomer.
  - (a) Saran (b) Carbowax.

### Q. 3 ANSWER THE FOLLOWING

- (i) Discuss why **nucleophilic** substitution reaction in **pyridine** is preferred at the 2- and 4-positions.
- (ii) Arrange the following compounds in increasing basicity order and give detail explanation of your answer.

(a) Imine (b) Amine (c) Nitrile.

OR

### Q. 3 ANSWER THE FOLLOWING

- (i) Why electrophilic substitution reaction in pyrrole occurs at carbon atom and not on hetero atom.
- (ii) Give the synthesis of 5,6-Benzoquinoline from 2-aminonaphthalene and glycerol by Skraup synthetic route.

## Q. 4 ANSWER THE FOLLOWING

(i) Molecular formula :  $C_4H_6O$ 

<b>CMR</b> ( $\delta$ , <b>ppm</b> ): (a) 3.4, Quartet (b) 50.8	, Triplet
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(c) 77.9, Singlet (d) 81.6, Singlet.

NMR (δ, ppm): (a) 2.0, 3H, Singlet (b) 1.8, 1H, Singlet (c) 4.1, 2H, Singlet.



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## (ii) Molecular formula : C<sub>9</sub>H<sub>10</sub>O<sub>2</sub>

NMR (δ, ppm): (a) 7.5, 4H, Quartet (b) 3.9, 3H, Singlet (c) 2.5, 3 H, Singlet .

**IR (Cm<sup>-1</sup>) :** 3000, 2900, 1670, 1598, 1500, 1450, 1375, 1258, 1171,1021, 833.

#### OR

## Q. 4 ANSWEP THE FOLLOWING

(i) Molecular formula : C<sub>9</sub>H<sub>10</sub>

IR (CM<sup>-1</sup>): 3100, 2950, 1650, 1600, 1500, 1450, 1375, 890, 760-770.

**NMR** ( $\delta$ , **ppm**): (a) 7.4, 5H, Complex (b) 5.35, 1H, Singlet

(c) 5.1, 1H, Singlet (d) 2.10, 3H, Singlet.

(ii) Discuss the proton-coupled and proton-decoupled CMR spectrum of sec-butyl bromide.

## Q. 5 ANSWER THE FOLLOWING

(i) What are dienes ? Just classify the following dienes into appropriate class.

(a) 2,4-hexadiene (b) Allene (c) 1,4-pentadiene.

(ii) What is coordination polymerization ? Explain the importance of Ziegler-Natta catalyst in coordination polymerization.

### OR

### Q. 5 ANSWER THE FOLLOWING

- (i) Discuss the addition of HBr to 1,3-butadiene at -80<sup>o</sup>C and at 40<sup>o</sup>C temperature with potential energy diagram.
- (ii) Give the distinguishing features of addition polymerization and condensation polymerization.

There is no short cut, except hard work with understanding to excel in examination.



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