## V.P. \& R. P. T. P. SCIENCE COLLEGE, VALLABH VIDYANAGAR

B. Sc. (Semester - V) Examination
$3^{\text {rd }}$ October 2013, Thursday
INORGANIC CHEMISTRY (US05CCHE03)
Time: 3.30 pm To 5.00 pm
Total Marks: 30
Q. 1 Answer the following MCQs:
i. Which of the following is the principle axis of rotation in $\mathrm{NH}_{3}$ molecule?
(a) $\mathrm{C}_{2}$
(b) $\mathrm{C}_{3}$
(c) $\mathrm{C}_{4}$
(d) $\mathrm{C}_{1}$
ii. Point group of $\mathrm{CH}_{4}$ molecule is $\qquad$ .
(a) $\mathrm{O}_{\mathrm{h}}$
(b) $\mathrm{T}_{\mathrm{d}}$
(c) $\mathrm{D}_{3} \mathrm{~h}$
(d) $\mathrm{D}_{6} \mathrm{~h}$
iii. Which is the best theory to understand coordinate bonding?
(a) M.O.Theory
(b) LFT
(c) CFT
(d) VBT
iv. Which type of substance or metal has higher susceptibility to magnetism?
(a) Metal ions
(b) Ferromagnetic
(c) Diamagnetic
(d) Antiferromagnetic
v. The term used for kinetic stability are $\qquad$ -
(a) labile \& inert
(b) labile \& stable
(c) labile \& unstable
(d) inert \& unstable
vi. In aqueous solution, the concentration of water is $\qquad$ .
(a) $[5.55 \mathrm{M}]$
(b) $[0.55 \mathrm{M}]$
(c) $[5.00 \mathrm{M}]$
(d) $[55.5 \mathrm{M}]$

## Q. 2 Answer the following short questions (Any three):

i. Define the terms: (a) symmetry operation (b) symmetry element
ii. Construct the multiplication table for $\mathrm{C}_{2 \mathrm{v}}$ point group.
iii. What is crystal field splitting?
iv. Explain in short high spin complex.
v. Give factors affecting stability of complexes.
vi. Define the terms: (a) substrate
(b) Anation reaction

Q. 3
a) Give an account of $D_{n}$ and $D_{n h}$ groups.
b) Prove that $\mathrm{Sn}^{\mathrm{n}}=\mathrm{E}$ giving proper example, where $\mathrm{n}=$ even number.

## OR

Q. 3
a) Prove that $\mathrm{C}_{3 v}$ is a non-abelian group giving sutiable example.
b) Give account of $\mathrm{C}_{\mathrm{n}}$ and $\mathrm{C}_{\mathrm{nv}}$ groups.

## Q. 4

a) Explain splitting of d-orbitals in octahedral complexes.
b) Write note on Jhan Teller effect.


## OR

Q. 4
a) Explain $\left[\mathrm{Ti}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{+3}$ is purple or violet in color.
b) Calculate in the unit of $\Delta_{0}$, the LFSE of $\mathrm{Fe}^{2+}(\mathrm{Z}=26)$ high spin ion in octahedral complex. (Given : $\Delta_{0}=10,400 \mathrm{~cm}^{-1}, \mathrm{P}=17,600 \mathrm{~cm}^{-1}$ )

## OR

Q. 5
a) Discuss the $\mathrm{S}_{\mathrm{N}} 1$ mechanism in ligand substitution reaction in octahedral complexes.
b) Discuss the continuous variation method for the determination of composition of the complex.

## OR

Q. 5
a) Discuss the acid hydrolysis reaction of six coordinated Co (III) ammine complexes.
b) What is trans effect? Discuss the electrostatic polarization theory for trans effect.

