# V.P. & R. P. T. P. SCIENCE COLLEGE, VALLABH VIDYANAGAR B. Sc. (Semester - V) Examination 3<sup>rd</sup> October 2013, Thursday INORGANIC CHEMISTRY (US05CCHE03) Time: 3.30 pm To 5.00 pm Total Marks: 30

Q.1 A i.	nswer the follo Which of the	o <mark>wing N</mark> followin	<b>ICQs:</b> ig is the princi	ple axis	of rotation in NH <sub>3</sub> mo	olecule?	(06)
	(a) C <sub>2</sub>	(b) C <sub>3</sub>	(c) C <sub>4</sub>		(d) C <sub>1</sub>		
ii.	Point group of CH <sub>4</sub> molecule is						
	(a) O <sub>h</sub>	(b) T <sub>d</sub>	(c) D <sub>3</sub>	h	(d) D <sub>6</sub> h		
iii.	Which is the best theory to understand coordinate bonding?						
	(a) M.O.Theo	ry	(b) LFT		(c) CFT	(d) VBT	
iv.	Which type of substance or metal has higher susceptibility to magnetism?						
	(a) Metal ions		(b) Ferromag	netic	(c) Diamagnetic	(d) Antiferro	magnetic
v.	The term used for kinetic stability are						
	(a) labile & in	iert	(b) labile & s	table	(c) labile & unstable	(d) inert & u	nstable
vi.	In aqueous solution, the concentration of water is						
	(a) [5.55M]		(b) [0.55M]		(c) [5.00M]	(d) [55.5M]	
Q.2 A i.	nswer the follo Define the ter	owing s ms: (a)	nort questions	s (Any th ration	n <b>ree):</b> (b) symmetry elemen	nt	(06)
ii.	Construct the multiplication table for $C_{2v}$ point group.						
iii.	What is crystal field splitting?						
iv.	Explain in short high spin complex.						
v.	Give factors affecting stability of complexes.						
vi.	Define the terms: (a) substrate (b) Anation reaction						/
Q.3 a) b)	Give an accour Prove that Sn <sup>n</sup>	int of D <sub>r</sub> = E givir	and D <sub>nh</sub> group ng proper examp	os. ole, where	e n= even number.		(03) (03)
OR							
Q.3							

# a) Prove that C<sub>3v</sub> is a non-abelian group giving sutiable example. (03) b) Give account of C<sub>n</sub> and C<sub>nv</sub> groups. (03)

# Q.4

- a) Explain splitting of d-orbitals in octahedral complexes.
- b) Write note on Jhan Teller effect.



OR

# Q.4

- a) Explain  $[Ti(H_2O)_6]^{+3}$  is purple or violet in color.
- (03)b) Calculate in the unit of  $\Delta_0$ , the LFSE of Fe<sup>2+</sup> (Z=26) high spin ion in octahedral complex. (Given :  $\Delta_0 = 10,400 \text{ cm}^{-1}$ , P = 17,600 cm<sup>-1</sup>) (03)

#### OR

# Q.5

- a) Discuss the S<sub>N</sub>1 mechanism in ligand substitution reaction in octahedral (03)complexes.
- b) Discuss the continuous variation method for the determination of composition (03)of the complex.

# OR

# Q.5

- a) Discuss the acid hydrolysis reaction of six coordinated Co (III) ammine (03)complexes.
- b) What is trans effect? Discuss the electrostatic polarization theory for trans effect. (03)